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Specific Activity and Radiation Hazard of Radionuclides in Wheat and Bean Produced Near Shazand, Iran

Monire Mohebian¹, Reza Pourimani^{1*}

1. Department of Physics, Faculty of Science, Arak University, Arak, Iran

ARTICLE INFO	A B S T R A C T
<i>Article type:</i> Original Paper	Introduction: Radionuclides found in foods are harmful to human health. Wheat and bean are among the most important food ingredients in the world. Therefore, this study aimed to determine the specific activity of actual redirection of the redirection of the redirection of the redirection.
Article history: Received: Oct 26, 2019 Accepted: Jan 02, 2020	<i>Material and Methods:</i> In order to determine the specific activity of radionuclides, the gamma-ray spectrometry method was used employing a high-purity germanium detector with a relative efficiency of 80%.
<i>Keywords:</i> Dosage ELCR Radiation Specific activity	Results: Our findings showed that the specific activity of the ²²⁶ Ra isotope of radium had the ranges <1.31- 5.27 and <1-5.06 Bq/kg for wheat and bean samples, respectively. Moreover, the specific activity of the ²³² Th isotope of thorium was in the range of not detected (ND)-4.09 and ND-3.62 Bq/kg with the mean values of 2.19 and 2.69 Bq/kg for wheat and bean samples, respectively. The specific activity of the ⁴⁰ K isotope of potassium was obtained as 103.19-168.94 and 129.22-568.98 Bq/kg with the mean values of 142.21 and 458.37 Bq/kg for wheat and bean samples, respectively. The annual effective dose for wheat and bean intake was 0.11-0.52 and 0.02-0.18 mSv, respectively. Furthermore, the mean of excess lifetime cancer risk for wheat and bean samples was calculated as 1.06×10^{-3} and 0.11×10^{-3} , respectively. The latter values are lower than the world average for bean samples. Conclusion: According to the results of this study, the radiological parameters of wheat were higher than the global average and reference value, which may be due to ash dispersion in this area. For bean, these parameters were lower than the mean value. As a result, it could be concluded that bean is not considered as a threat to consumer health.

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Introduction

Wheat is one of the important food components in the world and contributes to 19% of the energy and 20.8% of the protein worldwide. According to the reports of the Food and Agriculture Organization (2012), wheat makes up about half of the world's diet and its annual average per capita consumption in the world is 67 kg [1]. Furthermore, beans are a rich source of protein with an average consumption per capita of 50 and 9 kg in the world and Iran, respectively [1]. Approximately 13,400,000 metric tons of wheat were produced in Iran in 2018 and per capita consumption for Iran was estimated as 167.6 kg/y.

Natural radioactive is widespread throughout the earth in various geological forms, including clay, rocks, plants, food, and water [2-4]. The specific concentration of radioactive nuclei in the soil and plants depends on soil constituents and environmental pollution resulting from chemical fertilizers, oil, coal burning, and mineral extraction, such as uranium and thorium production [5].

Nuclear radiation is harmful to human health and one of its important sources is the inhalation of radon

gas and the consumption of radionuclides in food. The radiation coming from the disintegration of the ²³⁸U and ²³²Th series, as well as ⁴⁰K nuclei, are considered as the most important environmental radiation sources [6,7]. The average annual effective dose absorbed by people from the natural sources of ionizing radiation is estimated to be 2.4 mSv/y, of which 0.32 mSv/y comes from the ingestion of radionuclides [8].

In addition to natural nuclides, the ¹³⁷Cs radioactive nuclei occur in nature artificially. The source of this particular radioactive nucleus can be human activities in the field of energy production and military operations, namely nuclear weapons testing. Moreover, nuclear incidents, such as the Fukushima earthquake (2011) and the Chernobyl accident (1986) can contribute to this type of radiation [9].

Thorium, uranium, and potassium, depending on the biological effect, accumulate in certain organs of the body. The lungs and kidneys are known as the target organs for uranium storage. Thorium and potassium accumulate in the lungs, liver, bones, and muscles. This particular nucleus can cause

^{*} Corresponding Author: Tel: +98-8634173318; E-mail: r-pourimani@araku.ac.ir

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biochemical changes, including chromosomal mutations or cancer when sits in the organs [10]. With this background in mind, the purpose of this study was to investigate the specific activity of natural radionuclides in the wheat and bean produced in the vicinity of a petrochemical complex. Biological parameters, such as the average annual effective dose and cancer risk were calculated for the collected samples.

Materials and Methods

Sampling and Sample Preparation

Samples were collected from villages near Shazand Petrochemical Company, Iran. The samples were cleaned and then washed with double distilled water. Drying of the samples was carried out for 24 h at room temperature and 6 h in the oven of 80°C. Samples were milled and passed through a No. 40 mesh (0.635 mm) to form a fine powder (Figure 1). A total of 200 g of the prepared samples was packed in 300 cm³ polyethylene containers and sealed with silicone glue. Samples were kept in the laboratory for at least 50 days, which was necessary for the balance of activity between ²²⁶Ra and ²²²Rn according to Equation 1 [11].

$$t_m = \frac{\ln \left(\lambda_2 - \lambda_1\right)}{\frac{\lambda_2}{\lambda_1}} \tag{1}$$

Where t_m is the activity balance time, and λ_1 and λ_2 represent the decay constants of 226 Ra and 222 Rn radionuclides, respectively.

Spectroscopy and Spectrum Analysis

In order to determine the amount and type of radioactive nuclei, the gamma spectrometry method was utilized. High-Purity Germanium (HPGe) detector (GCD30195 BSI manufactured by Baltic Scientific Instruments company, LV-1005, Latvia) was used with a relative efficiency of 80%, a working voltage of 3000 V, and an energy resolution of 1.85 keV for the gamma line of 1332.52 keV due to ⁶⁰Co (E&G Ortec company, Tennessee, USA). The energy and efficiency calibrations were performed by standard sources containing ¹⁵²Eu, ²⁴¹Am, and ¹³⁷Cs radionuclides of known activity. The absolute detector-sample efficiency was calculated using Equation 2:

$$\varepsilon = \frac{G-B}{A \times BR(\%) \times T} \times 100$$
 (2)

Where ε indicates the energy efficiency of the gamma radiation of the detector-sample set. G and B represent the gross and background count under the peak of full energy, respectively. A denotes the radionuclide known activity BR is the gamma branching factor - Radiation intensity emitted by the radionuclide, and T represents the time of spectra acquisition [12, 13].

Specific Activity Measurement

According to the results of gamma-ray spectra, the specific activity value was measured using Equation 3 [11]:

$$A = \frac{G - B}{\varepsilon \times Br \times M \times T}$$
(3)

Where A denotes the specific activity of radionuclide in the sample. G, B, BR, and T are the same as the aforementioned equation and ε represents the energy efficiency for the gamma-ray of detector-sample setup. Moreover, m denotes the mass of samples in kg.

The activity concentration of ²²⁶Ra was calculated by applying the gamma-ray line of ²¹⁴Pb (351.93 keV) with the intensity of 37.6% and ²¹⁴Bi (609.31 keV) with the intensity of 46.1%. The gamma line of ²²⁸Ac (911.07 keV) with the intensity of 29% was used to determine the activity of ²³²Th. In addition, the activity concentrations of ⁴⁰K and ¹³⁷Cs were measured directly using their own gamma-rays of 1460.75 (10.7%) and 661.66 keV (85.12%), respectively [14,15]. The minimum detectable activity (MDA) of the detector set for all radionuclides was calculated using Equation 4 [14]. Values lower than MDA are unacceptable and for these MDA radionuclides listed in the tables with the sign <. and for values below the MDA, the MDA values are listed in Tables and marked with <

$$MDA = \frac{LLD \times 100}{\varepsilon(\%) \times Br \times T}$$
(4)

Where LLD is expressed as Equation 5:

$$LLD = 2.70 + 4.65 \times \sqrt{N}$$
(5)

Where LLD is low-level detection and N represents the count under the full-energy peak. The definitions of ϵ , Br, and t are similar to Equation 3.

Calculating DI_r, Ra_{eq}, AED, the threshold consumption rate of wheat and bean, and ELCR

Taking into account the annual consumption of wheat and bean per capita in Iran (i.e.; 167.63 and 9.5 kg, respectively), the daily intake of radionuclides (DI_r) is calculated by Equation 6 [16]:

$$\mathrm{DI}_{\mathrm{r}} = \frac{\mathrm{A}_{\mathrm{r}} \times \mathrm{C}_{\mathrm{ds}}}{365} \tag{6}$$

Where DI_r , A_r , and C_{ds} are daily intake of radionuclides (Bq/d), radionuclide concentration (Bq/kg), and annual food intake (kg/y), respectively. The total radioactivity of natural radionuclides, including ²²⁶Ra, ²³²Th, and ⁴⁰K in the samples can be defined as the equivalent of ²²⁶Ra radioactivity, which is calculated using equation 7 [8]. Ra _(eq)= A_{Ra} +1.43 A_{Th} +0.077 A_K (7)

Radium equivalent (Ra_{eq}) was defined based on the assumption that 10 Bq/kg ²²⁶Ra, 7 Bq/kg ²³²Th, and 130 Bq/kg ⁴⁰K induce the same dose. The annual effective dose (AED) for an adult as a result of consuming foods containing natural radionuclides is calculated using equation 8 [8]:

(8)

$$AED = CR \times \sum_{i}^{3} A_{i} \times CF_{i}$$

Where CR, A_i , and CF_i indicate annual consumption rate (kg/y), specific radionuclide activity (Bq/kg), and dose conversion factor (Sv/Bq), respectively (see Table 1). Equation 9 was used to determine the maximum annual allowable wheat and bean consumption, taking into account that the average AED due to radionuclides intake was estimated as 0.32 mSv/y [8].

$$CR = \frac{AEDE}{\sum_{i}^{3} A_{i} \times CF_{i}}$$
(9)

Where A_i and CF_i represent the same factors as equation 8.

The cancer risk associated with eating food contaminated with radioactive nuclei was calculated using equation 10 [17]:

$$ELCR = ML \times AED \times RF \tag{10}$$

Where ELCR is Excess lifetime cancer risk, ML denotes average lifetime (71.15 years), AED is annual effective dose is annual radionuclide intake, and RF indicates a nominal risk coefficient for the public, which is 4.1×10^{-2} Sv⁻¹ [18,19].

Table 2. Specific activities of ²²⁶Ra, ²³²Th, ^{40K}, and ¹³⁷Cs in wheat samples

Table 1. Dose conversion and risk factor [20]

Dose conversion and risk factor								
Nuclide	²²⁶ Ra	²³² Th	⁴⁰ K					
CF_i (Sv/Bq)	2.80E-07	2.30E-07	6.20E-09					

Results

In the present study, the specific activities of ²²⁶Ra, ²³²Th, and ⁴⁰K radionuclides were measured in 11 wheat and 11 bean samples collected from an area near the oil industry. The corresponding values with the means and MDAs are presented in tables 2 and 3. For all samples, the specific activity of ¹³⁷Cs, which is also shown in tables 2 and 3, was lower than the minimum detectable activity.

Radiological parameters, such as Ra_{eq} , AED due to wheat and bean consumption, annual consumption threshold, and ELCR are summarized in tables 4 and 5. The contribution of radionuclides to total AED and the calculated correlation coefficient between ELCR and total AED for all samples are shown in figures 2 and 3.

ID Sampla	Specific activity and MDA value (Bq/kg)									
ID Sample	²²⁶ Ra	MDA	²³² Th	MDA	40 K	MDA	¹³⁷ Cs	MDA		
w1	<mda< td=""><td>1.88</td><td>3.47±1.07</td><td>2.02</td><td>145.75 ± 8.18</td><td>9.72</td><td><mda< td=""><td>0.42</td></mda<></td></mda<>	1.88	3.47±1.07	2.02	145.75 ± 8.18	9.72	<mda< td=""><td>0.42</td></mda<>	0.42		
w2	1.5 ± 0.78	1.23	3.33 ± 1.18	1.74	148.68 ± 8.04	9.2	<mda< td=""><td>0.42</td></mda<>	0.42		
w3	4.43 ± 1.14	1.59	$3.97{\pm}1.31$	2.31	150.72 ± 9.1	11.53	<mda< td=""><td>0.52</td></mda<>	0.52		
w4	<mda< td=""><td>1.31</td><td>ND</td><td></td><td>103.19 ± 7.16</td><td>10</td><td><mda< td=""><td>0.44</td></mda<></td></mda<>	1.31	ND		103.19 ± 7.16	10	<mda< td=""><td>0.44</td></mda<>	0.44		
w5	$3.34{\pm}1.08$	1.47	4.09 ± 1.18	2.13	164.68 ± 8.9	10.17	<mda< td=""><td>0.15</td></mda<>	0.15		
w6	1.87 ± 0.86	1.26	ND		168.94 ± 8.55	9.06	<mda< td=""><td>0.4</td></mda<>	0.4		
w7	$3.23{\pm}1.06$	1.38	$3.14{\pm}1.08$	2.03	129.22±7.64	9.4	<mda< td=""><td>0.19</td></mda<>	0.19		
w8	3.31±1	1.35	$3.54{\pm}1.08$	1.98	150.07 ± 8.29	9.5	<mda< td=""><td>0.43</td></mda<>	0.43		
w9	4.62 ± 1.09	1.3	$2.57{\pm}1.23$	1.8	134.73±7.83	9.53	<mda< td=""><td>0.16</td></mda<>	0.16		
w10	5.27 ± 1.11	1.4	<mda< td=""><td>1.98</td><td>121.93±7.33</td><td>9.25</td><td><mda< td=""><td>0.16</td></mda<></td></mda<>	1.98	121.93±7.33	9.25	<mda< td=""><td>0.16</td></mda<>	0.16		
w11	2.95 ± 0.83	1.25	<mda< td=""><td>2.12</td><td>146.42 ± 8.09</td><td>9.45</td><td><mda< td=""><td>0.38</td></mda<></td></mda<>	2.12	146.42 ± 8.09	9.45	<mda< td=""><td>0.38</td></mda<>	0.38		
mean	2.77±0.81		2.19±0.74		142.21±8.1					

Table 3. Specific activities of ²²⁶Ra, ²³²Th, ⁴⁰K, and ¹³⁷Cs in bean samples

ID Samuela		Specific activity (Bq/kg)										
ID Sample	²²⁶ Ra	MDA	²³² Th	MDA	40 K	MDA	¹³⁷ Cs	MDA				
B1	<mda< td=""><td>1.03</td><td><mda< td=""><td>1.93</td><td>519.67±19.22</td><td>10.35</td><td><mda< td=""><td>0.36</td></mda<></td></mda<></td></mda<>	1.03	<mda< td=""><td>1.93</td><td>519.67±19.22</td><td>10.35</td><td><mda< td=""><td>0.36</td></mda<></td></mda<>	1.93	519.67±19.22	10.35	<mda< td=""><td>0.36</td></mda<>	0.36				
B2	$1.84{\pm}1.1$	1.35	3.62 ± 1.29	2.06	$482.87{\pm}18.58$	10.1	<mda< td=""><td>0.2</td></mda<>	0.2				
B3	2.08 ± 1	1.3	$3.54{\pm}1.07$	1.76	541.65±20.2	9.95	<mda< td=""><td>0.37</td></mda<>	0.37				
B4	5.06 ± 1.14	1.5	3.47±1.2	2.22	515.38±19.39	9.89	<mda< td=""><td>0.48</td></mda<>	0.48				
В 5	<mda< td=""><td>1.57</td><td>3.23 ± 1.22</td><td>1.8</td><td>$520.64{\pm}19.84$</td><td>11.89</td><td><mda< td=""><td>0.47</td></mda<></td></mda<>	1.57	3.23 ± 1.22	1.8	$520.64{\pm}19.84$	11.89	<mda< td=""><td>0.47</td></mda<>	0.47				
B 6	$2.84{\pm}1.08$	1.5	3.15±1.16	2.14	555.4±20.63	11.12	<mda< td=""><td>0.41</td></mda<>	0.41				
В 7	4.16±0.93	1.31	$2.86{\pm}1.1$	2.05	568.56±20.67	10.23	<mda< td=""><td>0.13</td></mda<>	0.13				
B 8	<mda< td=""><td>1</td><td>$3.49{\pm}1.21$</td><td>1.99</td><td>505.7±18.75</td><td>10.05</td><td><mda< td=""><td>0.13</td></mda<></td></mda<>	1	$3.49{\pm}1.21$	1.99	505.7±18.75	10.05	<mda< td=""><td>0.13</td></mda<>	0.13				
B 9	<mda< td=""><td>1.18</td><td>$2.86{\pm}1.01$</td><td>1.96</td><td>568.98±20.71</td><td>10.35</td><td><mda< td=""><td>0.24</td></mda<></td></mda<>	1.18	$2.86{\pm}1.01$	1.96	568.98±20.71	10.35	<mda< td=""><td>0.24</td></mda<>	0.24				
B 10	4.94 ± 0.98	1.3	3.4±1.09	1.8	134.03±7.6	9.13	<mda< td=""><td>0.14</td></mda<>	0.14				
B 11	$2.54{\pm}0.91$	1.3	ND^*		129.22±7.87	9.4	<mda< td=""><td>0.19</td></mda<>	0.19				
mean	2.13±0.65		2.69 ± 0.94		458.37±17.59							

ND: not detected



Table 4.	Radiological	parameters	due to	wheat	ingestion
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comula	Ra _{eq}	DI _r (Bq/d)				$C_{\rm rel}(l_{\rm rel}/{\rm rel})$	ELCR			
(Bq/kg	(Bq/kg)	²²⁶ Ra	²³² Th	⁴⁰ K	²²⁶ Ra	²³² Th	40 K	Total	Cr (kg/y)	(10^{-3})
W1	16.18	0	1.59	66.93	0	0.13	0.15	0.28	188.04	0.83
W2	17.71	0.69	1.53	68.27	0.07	0.13	0.15	0.35	151.82	1.03
W3	21.71	2.03	1.82	69.21	0.21	0.15	0.16	0.52	103.63	1.51
W4	7.95	0	0	47.38	0	0	0.11	0.11	500.17	0.31
W5	21.87	1.53	1.88	75.62	0.16	0.16	0.17	0.48	110.46	1.42
W6	14.88	0.86	0	77.57	0.09	0	0.18	0.27	203.69	0.77
W7	17.67	1.48	1.44	59.33	0.15	0.12	0.13	0.4	131.81	1.19
W8	19.93	1.51	1.63	68.91	0.16	0.14	0.16	0.46	119.79	1.31
W9	18.67	2.11	1.18	61.87	0.22	0.1	0.14	0.46	117.65	1.33
W10	14.66	2.41	0	55.99	0.25	0	0.13	0.38	143.4	1.09
W11	14.22	1.35	0	67.23	0.14	0	0.15	0.29	184.57	0.85
Min	7.95	0	0	47.38	0	0	0.11	0.11	103.63	0.31
Max	21.87	2.41	1.88	77.57	0.25	0.16	0.18	0.52	500.17	1.51
Mean	16.86	1.27	1.01	65.3	0.13	0.08	0.15	0.36	177.73	1.06

Table 5. Radiological parameters due to bean ingestion

Commla	Ra _{eq}	Ra _{eq} DI _r		AED (mSv/y)				$C_{\rm m}$ (leg/g)	ELCR	
(Bo	(Bq/kg)	²²⁶ Ra	²³² Th	⁴⁰ K	²²⁶ Ra	²³² Th	40 K	Total	Cr (kg/y)	(10^{-3})
bs1	40.01	0	0	238.62	0	0	0.03	0.03	99.32	0.09
bs2	44.2	0.84	1.66	221.72	0	0.14	0.03	0.17	73.71	0.12
bs3	48.85	0.95	1.63	248.71	0.01	0.14	0.03	0.18	67.30	0.13
bs4	49.71	2.32	1.59	236.65	0.01	0.13	0.03	0.17	59.15	0.15
bs5	44.71	0	1.48	239.07	0	0.12	0.03	0.15	80.59	0.11
bs6	50.11	1.3	1.45	255.03	0.01	0.12	0.03	0.16	64.47	0.14
bs7	52.03	1.9	1.31	261.07	0.01	0.11	0.03	0.15	59.84	0.15
bs8	43.93	0	1.6	232.21	0	0.13	0.03	0.16	81.26	0.11
bs9	47.9	0	1.31	261.26	0	0.11	0.03	0.14	76.45	0.12
bs10	20.12	2.26	1.56	61.54	0.01	0.13	0.01	0.15	106.8	0.08
bs11	12.49	1.16	0	59.33	0.01	0	0.01	0.02	211.59	0.04
min	12.49	0	0	59.33	0	0	0.01	0.02	59.15	0.04
max	52.03	2.32	1.66	261.26	0.01	0.14	0.03	0.18	211.59	0.15
mean	41.28	0.98	1.24	210.47	0.01	0.10	0.03	0.14	89.13	0.11

Discussion

The specific activity of ²²⁶Ra had the ranges of <1.31-5.27 and <1-5.06 (Bq/kg) for wheat and bean samples, respectively. In addition, the values for ²³²Th were in the ranges of ND-4.09 and ND-3.62 (Bq/kg) with the mean of 2.19 and 2.69 (Bq/kg) for wheat and bean samples, respectively. The specific activity of ⁴⁰K was obtained as 103.19-68.94 and 129.22-568.98 (Bq/kg) with the mean values of 142.21 and 458.37 (Bq/kg) for wheat and bean samples, respectively. The latter result shows that bean is more enriched in terms of mineral potassium content.

Cesium level in all samples was less than the minimum detectable activity indicating that cesium in the soil is associated with larger molecules and cannot be easily absorbed by the root of a plant. As demonstrated in tables 4 and 5, the calculated DI_rs due to the consumption of wheat samples were reported as 0-2.41, 0–1.88, and 47.38–77.57 Bq for ²²⁶Ra, ²³²Th, and ⁴⁰K, respectively. Furthermore, the DI_rs resulting from bean consumption were found as 0-2.32, 0-1.66,

and 59.33-261.26 Bq for 226 Ra, 232 Th, and 40 K, respectively.

The Ra_{eq} value for wheat and bean samples was in the range of 7.95-21.87 with a mean of 16.84 Bq/kg and the range of 12.49-52.03 with the mean of 41.28 (Bq/kg), respectively. The mentioned finding reveals that bean has more radioactivity content, compared to wheat. The consumption-induced AEDs for wheat and bean samples were within the range of 0.11-0.52 with a mean of 0.36 and the range of 0.02-0.18 with the mean value of 0.14 mSv/y, respectively.

These results show that for the wheat cultivated in this region, the average ingestion dose is higher than the world average of 0.32 mSv/y. However, it lays within the acceptable AED of 1 mSv/y for the public [7]. Acceptable consumption rate per capita for wheat was in the range of 103.63-500.17 with the mean value of 177.73 kg/y that for most of the samples is lower than the annual consumption rate in Iran (167.63 kg/y).

Natural radionuclides in the wheat sample of this investigation were more than other countries and other

parts of Iran, which could be attributed to environmental pollution due to the waste of oil burning that disperses to farms in the form of fly ash. This indicates that the wheat cultivated in this area is mostly contaminated by radioactive nuclei. Therefore, constant monitoring is required when this wheat, which is associated with health risks to the population is consumed. On the other hand, bean consumption does not threaten public health because the mean AED value of bean is definitely lower than the world average (0.05<0.32) [8].

The contribution of each radioactive element to the AED was calculated as shown in Figure 2 for both wheat and bean. In wheat samples, ⁴⁰K had the main share of 41% in the dose, while ²³²Th series had the largest share of 77% in bean seeds. The mean values for wheat and bean were 1.06×10^{-3} and 0.11×10^{-3} , respectively. Consequently, the ELCR of wheat was higher than the world average of 0.29×10^{-3} and was equal to the maximum acceptable limit of 10^{-3} , respectively [8].

Figure 3 shows a good correlation between ELCR and total AED in all samples. Correlation coefficient values (r^2) were obtained as 0.99 and 0.6 for wheat and bean samples, respectively. Table 6 lists the reported values of specific activity for natural radionuclides in wheat and bean in some countries. Comparison of the specific activity of radionuclides in this study with the data from other reports reveals a higher amount of ²²⁶Ra, ²³²Th, and Ra_{eq} in wheat and bean samples. However, in the case of potassium was in the same range except for Tanzania.

The analyzed data showed that the average radiological parameters of wheat were higher than the world average and acceptable limits, which indicates the impact of environmental pollution on the wheat produced in this region. Regarding bean, these parameters were found to be within the acceptable interval. As a result, this group of plants does not pose a radiological hazard to humans.



Figure 2. Share of radionuclides in the average annual effective dose in wheat and bean samples



Figure 3. Correlation between AED and ELCR in wheat (left) and bean (right) samples

Country	Specific	Specific Activity (Bq/kg)			Average Annual Effective Dose (mSv/y)				CR	Reference
Country	²²⁶ Ra	²³² Th	⁴⁰ K	raad (Dd/ng)	²²⁶ Ra	²³² Th	⁴⁰ K	Total	(kg/y)	Reference
Wheat										
France	0.57	< 0.03	146.3	11.8						[23]
Iraq (Karbala)	<1.46	<1.38	180.54	13.9	*	*	*	1	*	[13]
Iran (Ilam)	1.67	0.5	91.73	8.15	*	*	*	5.28		[21]
Iran (Shazand)	2.77	2.19	142.21	16.86	0.13	0.25	0.15	0.53	177.73	Present study
Bean										
Tanzania	0.19	0.17	38.22	3.38	8.55	12.5	45.1	0.06	200	[22]
Brazil	1.43	0	434	34.85	*	*	*	*	*	[23]
Iran (Arak)					0.03	2.05	0.55	0.55		[24]
Iran (Shazand)	2.13	2.69	458.37	41.27	0.01	0.01	0.03	0.05	89.13	Present study
world average								290		[8]
Cereals (wheat, cor	n, and bean)									
Sudan (Khartoum)	0.20	0.63	230.85	18.88	*	*	*	0.57	392.2	[25]
Sudan (South Korde	ofan) 4.31	3.36	138.05	19.74	*	*	*	1.27	392.2	[25]
Sudan (Tokar)	0.88	1.62	323.96	28.14	*	*	*	0.33	392.2	[25]
World mean					*	*	200	290		[8]

Table 6. Comparison of the specific radionuclide activity, average annual effective dose, and consumption rate of wheat and bean samples with other studies in the world

Conclusion

Natural radioactivity values (²²⁶Ra, ²³²Th, and ⁴⁰K) in wheat and bean were measured by gamma-ray spectrometry method using HPGe detector. Results of the current study indicated that the ²²⁶Ra and ²³²Th contents of wheat and bean were higher than the other reports except for Sudan. The ELCR for wheat consumption was higher than the world average and was equal to the maximum acceptable value. On the other hand, the corresponding value for bean was observed to be lower than the world average.

Therefore, it seems that the emission of pollutants from the combustion chimneys of petroleum fuels led to the increased uptake of radionuclides by wheat and bean. Consequently, it is necessary to investigate these changes in the future studies. Moreover, our results demonstrated that radiological parameters did not exceed the reference values in this stage and do not pose a radiological threat to human health.

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